AMENDMENTS TO THE CLAIMS

Claim 1 (Original) A process for generating $^1O_2$ which comprises treating an Sn(II) salt of the formula

$$SnX_n(I)$$

in which X is an anion from the group consisting of trifluoromethanesulfonate, acetate, formate, oxalate, lactate, malonate, malate, tartrate, citrate, orthophosphate, sulfate, chloride, perchlorate and n is 1 or 2, in an organic solvent at a temperature of from -80°C to 20°C with 1 to 2 mol of ozone per mole of tin compound, and using the $^1O_2$ which forms directly for the oxidation of organic substrates which react with $^1O_2$.

Claim 2 (Original) The process as claimed in claim 1, wherein the Sn(II) salt used is tin(II) trifluoromethanesulfonate or tin(II) acetate.

Claim 3 (Original) The process as claimed in claim 1, wherein the organic solvent used is ethyl acetate, butyl acetate, methanol, ethanol, dichloromethane or acetic acid.

Claim 4 (Original) The process as claimed in claim 1, wherein the reaction temperature is -80°C to -5°C.

Claim 5 (Original) The process as claimed in claim 1, wherein one equivalent of ozone is used.

Claim 6 (Original) The process as claimed in claim 1, wherein a solution of an organic substrate which reacts with $^1O_2$ is metered in during the reaction of the Sn(II) salt with ozone.

Claim 7 (Original) The process as claimed in claim 1, wherein a solution of an organic substrate which reacts with $^1O_2$ is metered in after the reaction of the Sn(II) salt with ozone, following removal of any excess ozone.

Claim 8 (Currently Amended) The process as claimed in claim 6 ~~or 7~~, wherein the solvent used for the substrate is ethyl acetate, butyl acetate, methanol, ethanol, dichloromethane or acetic acid.

Claim 9 (New) The process as claimed in claim 7, wherein the solvent used for the substrate is ethyl acetate, butyl acetate, methanol, ethanol, dichloromethane or acetic acid.